

Phenolic Compounds from *Nymphaea odorata*

Zhizhen Zhang,[†] Hala N. ElSohly,^{*,†} Xing-Cong Li,[†] Shabana I. Khan,[†] Sheldon E. Broedel, Jr.,[‡] Robert E. Rauli,[‡] Ronald L. Cihlar,[§] Charles Burandt,[†] and Larry A. Walker^{*,†,⊥}

National Center for Natural Products Research, Research Institute of Pharmaceutical Sciences, and Department of Pharmacology, School of Pharmacy, University of Mississippi, University, Mississippi 38677, Dorlin Pharmaceuticals, Baltimore, Maryland 21227, and Department of Microbiology and Immunology, Georgetown University, Washington, D.C. 20057

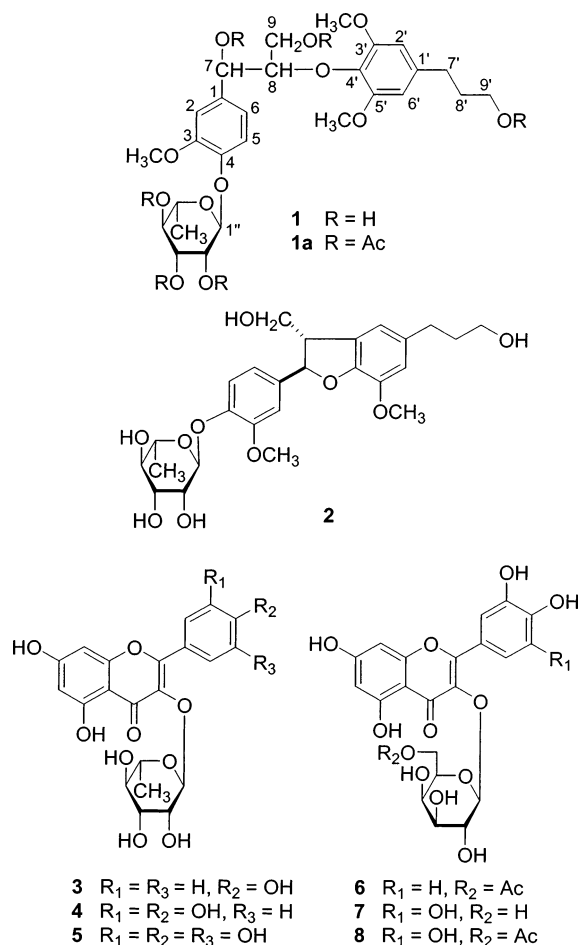
Received September 11, 2002

Assay-guided fractionation of the ethanol extract of *Nymphaea odorata* resulted in the identification of two lignans, one new (**1**) and one known (**2**), together with six known flavonol glycosides (**3**–**8**). The structures of **1**–**8** were established by spectroscopic analysis as nymphaeoside A (**1**), icaricide E₄ (**2**), kaempferol 3-*O*- α -L-rhamnopyranoside (afzelin, **3**), quercetin 3-*O*- α -L-rhamnopyranoside (**4**), myricetin 3-*O*- α -L-rhamnopyranoside (myricitrin, **5**), quercetin 3-*O*-(6''-*O*-acetyl)- β -D-galactopyranoside (**6**), myricetin 3-*O*- β -D-galactopyranoside (**7**), and myricetin 3-*O*-(6''-*O*-acetyl)- β -D-galactopyranoside (**8**). Compounds **3**, **4**, and **7** showed marginal inhibitory effect against fatty acid synthase with IC₅₀ values of 45, 50, and 25 μ g/mL, respectively.

Fatty acid synthase (FAS) is an enzyme essential to the infective process in *Candida albicans*.^{1,2} Early work on the structure–activity relationship among a series of structurally related FAS inhibitors showed that the fungal and human enzymes could be differentially inhibited.³ This suggested that with appropriate inhibitors FAS could be used as a therapeutic target. In the course of our continued efforts of searching for fatty acid synthase (FAS) inhibitors from natural sources,⁴ we investigated an active ethanolic extract of *Nymphaea odorata*.

Nymphaea odorata Ait. (Nymphaeaceae), or American white water lily, is a herbaceous hydrophyte native to the southeastern United States. Previous work on this plant resulted in the isolation of five compounds with plant growth-inhibitory properties.⁵ Bioactivity-guided fractionation of the leaves of *N. odorata* resulted in the isolation of eight constituents (**1**–**8**). This work describes the isolation, structure elucidation, and biological activity of these compounds.

Compound **1** was obtained as a colorless amorphous solid. Its molecular formula of C₂₇H₃₈O₁₂ was determined by HRESIMS and indicated nine degrees of unsaturation. The ¹³C NMR spectrum of **1** displayed 27 signals, of which 21 were assigned to the aglycone moiety including two aromatic rings, three methoxy groups, one hydroxypropyl group, and one 1,3-propanediol group. The remaining six signals corresponded to a deoxyhexose. The ¹H NMR spectrum of **1** displayed signals for a 1,3,4-trisubstituted aromatic ring [δ 7.11 (1H, br s, H-2), δ 7.08 (1H, d, *J* = 8.0 Hz, H-5), δ 6.93 (1H, d, *J* = 8.0 Hz, H-6), δ 3.85 (3H, s, OCH₃-3)], a 1',3',4',5'-tetrasubstituted aromatic ring [δ 6.57 (2H, s, H-2',6'), δ 3.83 (6H, s, OCH₃-3',5')], a 1,2,3-trioxygenated propyl function [δ 4.99 (1H, d, *J* = 5.2 Hz, H-7), δ 4.22 (1H, m, H-8), δ 3.95 (1H, m, H-9b), δ 3.57 (1H, m, H-9a)], and a hydroxypropyl group [δ 2.65 (2H, t, *J* = 8.0 Hz, H-7'), δ 1.82 (2H, m, H-8'), δ 3.57 (2H, m, H-9')].



These data suggested that the aglycone of **1** is a lignan, identified as 1-[4-hydroxy-3-methoxyphenyl]-2-[4-(3-hydroxypropyl)-2,6-dimethoxyphenoxy]-1,3-propanediol,^{6,7} and these inferences were confirmed by HMQC and HMBC correlations (Figure 1). Acid hydrolysis of **1** afforded rhamnose, which was identified by co-TLC with an authentic sample. The characteristic NMR signals of δ _H 5.38 (1H, br s, H-1'') and δ _H 1.26 (3H, d, *J* = 5.9 Hz, H-6''), as well as their protonated carbon signals at δ _C 101.4 (C-1'')

* To whom correspondence should be addressed. (H.N.E.) Tel: 662-915-7610. Fax: 662-915-7989. E-mail: helsohly@olemiss.edu. (L.A.W.) Tel: 662-915-1005. Fax: 662-915-1006. E-mail: lwalker@olemiss.edu.

[†] National Center for Natural Products Research, University of Mississippi.

[‡] Dorlin Pharmaceuticals.

[§] Georgetown University.

[⊥] Department of Pharmacology, University of Mississippi.

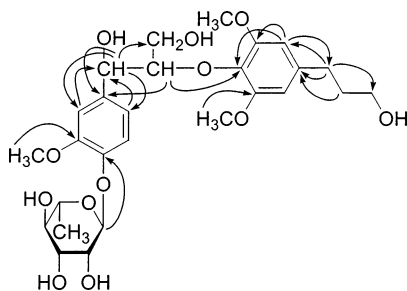


Figure 1. Key HMBC (H to C) correlations of **1**.

and δ_C 18.8 (C-6'') also indicated the presence of a rhamnose moiety. A HMBC correlation between H-1'' (δ 5.38) of rhamnose and C-4 (δ 146.0) of aglycone established that the rhamnose was attached to C-4 of aglycone. The α -pyranose form of rhamnose in **1** was confirmed^{8,9} by a set of carbon signals at δ 101.4, 72.2, 72.0, 73.8, 70.7 and 18.8, which correlated with δ 5.38 (br s), 4.12 (br s), 3.95 (m), 3.48 (t, $J = 9.5$ Hz), 3.91 (m), and 1.26 (d, $J = 5.9$ Hz), respectively, in the HMQC spectrum. The stereochemistry for C-7 and C-8 was shown to be *erythro*⁶ from the chemical shift value at δ 6.00 (1H, d) and the J coupling constant value (4.7 Hz) of H-7 in **1a**, the corresponding acetate derivative of **1**. HMQC and HMBC analyses allowed for the complete assignments of the ¹H and ¹³C signals of **1** (1-[4-*O*-(α -L-rhamnopyranosyl)-3-methoxyphenyl]-2-[4-(3-hydroxypropyl)-2,6-dimethoxyphenoxy]-1,3-propanediol), a new natural product named nymphaeoside A.

The known compounds were identified by comparison of their spectral data with reported values as icaraside E₄ (**2**),¹⁰ kaempferol 3-*O*- α -L-rhamnopyranoside (afzelin, **3**),¹¹ quercetin 3-*O*- α -L-rhamnopyranoside (**4**),¹² myricetin 3-*O*- α -L-rhamnopyranoside (myricitrin, **5**),¹³ quercetin 3-*O*-(6''-*O*-acetyl)- β -D-galactopyranoside (**6**),¹⁴ myricetin 3-*O*- β -D-galactopyranoside (**7**),¹⁴ and myricetin 3-*O*-(6''-*O*-acetyl)- β -D-galactopyranoside (**8**).¹⁴ This is the first report of the isolation of **2–8** from this plant.

Compounds **1–8** were evaluated in the fatty acid synthase (FAS) inhibition assay.⁴ The results showed compounds **3**, **4**, and **7** had weak inhibitory effects against FAS with IC₅₀ values of 45, 50, and 25 μ g/mL, respectively. Cerulenin¹⁵ was used as a positive control (IC₅₀ of 0.19 μ g/mL).

Experimental Section

General Experimental Procedures. Optical rotations were determined on a JASCO DIP-370 digital polarimeter. UV spectra were recorded on a Hewlett-Packard 8435 spectrometer. IR spectra were obtained on an ATI Mattson Genesis Series FTIR spectrometer. The NMR spectra were recorded on a Bruker Avance DRX-400 spectrometer operating at 400 MHz for ¹H and 100 MHz for ¹³C. 2D NMR spectra were measured on a Bruker Avance DRX-500 operating at 500 MHz using standard pulse programs and acquisition parameters. HRESIMS were measured on a Bruker-Magnex BioAPEX 30es ion cyclotron high-resolution HPLC-FT spectrometer by direct injection into an electrospray interface. Si gel (40 μ m, J. T. Baker) and RP Si gel (RP-18, 40 μ m, J. T. Baker) were used for low-pressure chromatography. HPLC was performed using an ODS column (Phenomenex, Prodigy ODS prep, 21.2 mm i.d. \times 250 mm, 10 μ m). TLC was performed on Si gel 60 F₂₅₄ (EM Science) using CHCl₃/MeOH (4:1, solvent A), toluene/EtOAc/MeOH (4:1:1, solvent B), and CHCl₃/EtOAc (6:1, solvent C) or reversed-phase KC₁₈ F Si gel 60 (Whatman) using MeOH/H₂O (70:30, solvent D). The detailed procedures for the bioassays are described in a previous paper.⁴

Plant Material. The plant material (leaves) was collected in Florida in July 1996 and identified by Dr. Charles Burandt. A voucher specimen is on deposit at the National Center for Natural Products Research, The University of Mississippi (voucher # BUR 190796 1A).

Extraction and Isolation. The dried and powdered (60 mesh) plant material (400 g) was percolated with 95% EtOH (3000 mL \times 3). The ethanolic extract was evaporated to dryness (50.1 g, IC₅₀ 25 μ g/mL). Part of the EtOH extract (33.0 g) was chromatographed over a silica gel column (300 g) eluting with CHCl₃/MeOH (9:1, 4:1, 1:1, 1:4, and 0:1, each 1000 mL). On the basis of TLC analysis, nine combined fractions were obtained: A (5.19 g), B (0.70 g), C (0.70 g), D (2.17 g), E (8.09 g), F (3.68 g), G (5.28 g), H (4.10 g), and I (2.25 g). Part of fraction B (0.66 g) was rechromatographed over a silica gel column (120 g) using CHCl₃/MeOH (7:1 and 4:1, each 1500 mL) to afford five fractions: B₁ (235.2 mg), B₂ (250 mg), B₃ (9 mg), B₄ (50.0 mg), and B₅ (60 mg). Part of B₂ (100 mg) was further applied onto a ODS column (10 g) washing with MeOH/H₂O (60:40) and then MeOH. The aqueous MeOH fraction (20 mg) was purified by HPLC (MeOH/H₂O, 60:40, 3 mL/min, UV 276 nm) to yield **1** (10.6 mg, t_R 22.5 min). Part of B₄ (40 mg) was separated using an ODS column (10 g) and washing with MeOH/H₂O (0:100, 50:50, and 80:20, each 200 mL). The 50% MeOH fraction (22 mg) was refractionated using HPLC (MeOH/H₂O, 50:50) to afford **2** (9.5 mg, t_R 21 min), **3** (2.1 mg, t_R 29.4 min), and **6** (2.6 mg, t_R 26 min).

Part of the FAS inhibitory fraction E (6.0 g) was chromatographed over a ODS column (200 g) eluting with MeOH/H₂O mixtures (100:0, 50:50, and 0:100, each 1000 mL) to give fractions E₁ (2.50 g), E₂ (1.48 g), E₃ (1.51 g), and E₄ (0.50 g). Fraction E₃ was further purified using HPLC (MeOH/H₂O, 50:50, 3 mL/min, UV 276 nm) to afford **4** (2.6 mg, t_R 54.1 min), **5** (8.0 mg, t_R 36.5 min), **7** (3.0 mg, t_R 33.0 min), and **8** (6.0 mg, t_R 40.5 min). Fractions F–I, E₁, and E₂, rich in very polar compounds (possibly tannins), were not further investigated.

Nymphaeoside (1): colorless amorphous powder; $[\alpha]_D^{25}$ -23.1° (c 0.5, MeOH); UV (MeOH) λ_{max} (log ϵ) 214 (4.05), 230 (sh), 276 (3.17) nm; IR (KBr) ν_{max} 3401, 1587, 1504, 1456, 1417, 1220, 1119, 1057, 1019, 973 cm^{-1} ; ¹H NMR (CD₃OD) δ 1.26 (3H, d, $J = 5.9$ Hz, H-6''), 1.82 (2H, m, H-8), 2.65 (2H, t, $J = 8.0$ Hz, H-7), 3.48 (1H, t, $J = 9.5$ Hz, H-4''), 3.57 (4H, m, H-9, 9'), 3.83 (6H, s, OCH₃-3', 5'), 3.85 (3H, s, OCH₃-3), 3.91 (1H, m, H-5''), 3.95 (1H, m, H-3''), 4.12 (1H, br s, H-2''), 4.22 (1H, m, H-8), 4.99 (1H, d, $J = 5.2$ Hz, H-7), 5.38 (1H, br s, H-1''), 6.57 (2H, br s, H-2', 6'), 6.93 (1H, d, $J = 8.0$ Hz, H-6), 7.08 (1H, d, $J = 8.0$ Hz, H-5), 7.11 (1H, br s, H-2); ¹³C NMR (CD₃OD) δ 154.3 (C-3', 5'), 151.6 (C-3), 146.0 (C-4), 139.9 (C-1), 138.0 (C-1'), 134.6 (C-4'), 120.4 (C-6), 119.2 (C-5), 112.4 (C-2), 106.8 (C-2', 6'), 101.4 (C-1''), 87.2 (C-8), 73.8 (C-4'), 73.7 (C-7), 72.2 (C-2''), 72.0 (C-3''), 70.7 (C-5''), 62.1 (C-9), 61.4 (C-9), 56.6 (OCH₃-3', 5'), 56.4 (OCH₃-3), 35.4 (C-8'), 33.4 (C-7'), 18.8 (C-6''); HRESIMS m/z 572.2698 [M + NH₄]⁺, 577.2258 [M + Na]⁺ (calcd for C₂₇H₃₈O₁₂, 572.2701 [M + NH₄]⁺, 577.2255 [M + Na]⁺).

Acetylation of 1. Compound **1** (6 mg) was dissolved in Ac₂O/pyridine (1:2, 3 mL), and the mixture was allowed to stand at room temperature for 24 h. After addition of 10 mL of H₂O, the resulting mixture was applied onto a low-pressure ODS column (5 g) washing with H₂O (50 mL) and then MeOH (30 mL). The MeOH fraction was evaporated to give **1a** (5 mg).

1a: colorless amorphous powder; $[\alpha]_D^{25}$ -35.4° (c 0.5, MeOH); UV (MeOH) λ_{max} (log ϵ) 214 (4.18), 230 (sh), 276 (3.14) nm; IR (KBr) ν_{max} 2940, 1744, 1589, 1508, 1460, 1369, 1220, 1124, 1039 cm^{-1} ; ¹H NMR (CDCl₃) δ 1.16 (3H, d, $J = 5.9$ Hz, H-6''), 1.88 (2H, m, H-8'), 1.95 (3H, s, -COCH₃), 1.99 (3H, s, -COCH₃), 2.04 (6H, s, -COCH₃ \times 2), 2.11 (3H, s, -COCH₃), 2.19 (3H, s, -COCH₃), 2.55 (2H, t, $J = 7.1$ Hz, H-7'), 3.70 (6H, s, OCH₃-3', 5'), 3.80 (3H, s, OCH₃-3), 4.05 (2H, m, H-9'), 4.12 (1H, m, H-5''), 4.20 (1H, m, H-9a), 4.39 (1H, m, H-9b), 4.55 (1H, br t, H-8), 5.07 (1H, t, $J = 9.8$ Hz, H-4''), 5.31 (1H, br s, H-1''), 5.49 (1H, br s, H-2''), 5.53 (1H, m, H-3''), 6.00 (1H, d, $J = 4.7$ Hz, H-7), 6.32 (2H, s, H-2', 6'), 6.82 (1H, d, $J = 8.0$ Hz, H-6), 6.92 (1H, br s, H-2), 6.98 (1H, d, $J = 8.0$ Hz, H-5); ¹³C NMR (CDCl₃) δ 171.5 (C=O), 171.3 (C=O \times 2), 170.4 (C=O \times

2), 170.1 (C=O), 153.4 (C-3',5'), 150.7 (C-3), 145.1 (C-4), 137.8 (C-1'), 133.6 (C-1)^a, 133.5 (C-4)^a, 119.8 (C-6), 118.7 (C-5), 112.0 (C-2), 105.6 (C-2', 6'), 97.7 (C-1''), 81.1 (C-8), 74.5 (C-7), 71.4 (C-4''), 70.1 (C-2'')^b, 69.3 (C-3'')^b, 67.7 (C-5''), 64.1 (C-9'), 63.1 (C-9), 56.3 (OCH₃-3, 3', 5'), 32.9 (C-7'), 30.5 (C-8'), 21.4 (-COCH₃), 21.3 (-COCH₃), 21.2 (-COCH₃ × 2), 21.1 (-COCH₃), 20.7 (-COCH₃), 17.7 (C-6''); assignments were based on COSY, HMQC, and HMBC spectra (^{a,b}The signals may be interchangeable); HRESIMS *m/z* 829.2871 [M + Na]⁺ (calcd for C₃₉H₅₀O₁₈, 829.2889 [M + Na]⁺).

Acid Hydrolysis of 1. Compound **1** and authentic rhamnose were spotted on a silica gel TLC plate and hydrolyzed in situ by exposure to HCl vapor at 70 °C for 25 min. The TLC plate was then developed with CHCl₃/MeOH/AcOH/H₂O (14:6:2:1) and sprayed with 10% H₂SO₄ for detection. Rhamnose was detected with an *R_f* value of 0.2.

Acknowledgment. The authors thank Dr. Chuck Dunbar for HRESIMS analyses, and Mr. Frank Wiggers for running gradient HMQC and HMBC NMR (500 MHz) experiments. This work was supported in part by the United States Department of Agriculture, Agricultural Research Service Specific Cooperative Agreement No. 58-6408-2-0009, and by NCDDG Grant #5 RO1 CA 88456-02.

Supporting Information Available: ¹H and ¹³C NMR data of **2–8**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References and Notes

- Zhao, X. J.; McElhaney-Feser, G. E.; Bowen, W. H.; Cole, M. F.; Broedel, S. E., Jr.; Cihlar, R. L. *Microbiology* **1996**, *142*, 2509–2514.
- Zhao, X. J.; McElhaney-Feser, G. E.; Sheridan, M. J.; Broedel, S. E., Jr.; Cihlar, R. L. *Infect. Immun.* **1997**, *65*, 829–832.
- Broedel, S. E., Jr. *Recent Res. Dev. Antimicrob. Agents Chemother.* **1996**, *1*, 25–33.
- Li, X.-C.; Joshi, A. S.; ElSohly, H. N.; Khan, S. I.; Jacob, M. R.; Zhang, Z. Z.; Khan, I. A.; Ferreira, D.; Walker, L. A.; Broedel, S. E.; Raulli, R. E.; Cihlar, R. L. *J. Nat. Prod.* **2002**, *65*, 1909–1914.
- Elakovich, S. D.; Spence, S.; Yang, J. Book of Abstracts, 214th ACS National Meeting, Las Vegas, NV, September 7–11, 1997; AGRO-036; American Chemical Society: Washington, DC.
- Hwang, B. H.; Sakakibara, A.; Miki, K. *Holzforchung* **1981**, *35*, 229–232.
- Hauteville, M.; Lundquist, K.; von Unge, S. *Acta Chem. Scand. B* **1986**, *40*, 31–35.
- Kasai, R.; Okihara, M.; Asakawa, J.; Mizutani, K.; Tanaka, O. *Tetrahedron* **1979**, *35*, 1427–1432.
- Alaoui, A.; Charrouf, Z.; Souflaoui, M.; Carbone, V.; Malorni, A.; Pizza, C.; Piacente, S. *J. Agric. Food Chem.* **2002**, *50*, 4600–4603.
- Miyase, T.; Ueno, A.; Takizawa, N.; Kobayashi, H.; Oguchi, H. *Phytochemistry* **1989**, *28*, 3483–3485.
- Matthes, H. W. D.; Luu, B.; Ourisson, G. *Phytochemistry* **1980**, *19*, 2643–2650.
- Markham, K. R.; Ternai, B.; Stanley, R.; Geiger, H.; Mabry, T. J. *Tetrahedron* **1978**, *34*, 1389–1397.
- Addae-Mensah, I.; Achenbach, H. *Phytochemistry* **1985**, *24*, 1817–1819.
- Foo, L. Y.; Lu, Y.; Molan, A. L.; Woodfield, D. R.; McNabb, W. C. *Phytochemistry* **2000**, *54*, 539–548.
- Price, A. C.; Choi, K.-H.; Heath, R. J.; Li, Z.; White, S. W.; Rocks, C. O. *J. Biol. Chem.* **2001**, *276*, 6551–6559.

NP020442J